



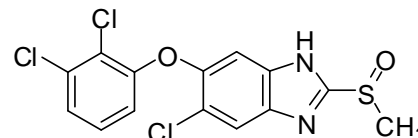
CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

NMIA P1685: Triclabendazole sulfoxide

Report ID: P1685.2022.01

Chemical Formula: C₁₄H₉Cl₃N₂O₂S

Molecular Weight: 375.7 g/mol



Certified value

Batch No.	CAS No.	Purity (mass fraction)
01-AV-06	100648-13-3	98.5 ± 1.0%

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit ($k = 2$).

IUPAC name: 6-Chloro-5-(2,3-dichlorophenoxy)-2-(methylsulfinyl)-1H-benzimidazole.

Expiration of certification: The property values are valid till 1 February 2027, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

Description: White crystalline solid sourced from an external supplier, and certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

Intended use: This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

Recommended storage: When not in use this material should be stored at or below 4 °C in a closed container in a dry, dark area.

Metrological traceability: The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. In the mass balance approach all impurities are quantified as a mass fraction and subtracted from 100%.

Stability: This material has demonstrated stability over a minimum period of 5 years. The measurement uncertainty at the 95% confidence interval includes a stability component which has been estimated from annual stability trials. This material is susceptible to oxidation to triclabendazole sulfone when taken into solution and exposed to air. Standard solutions prepared from this material should be stored out of direct light at 4 °C and monitored regularly for oxidation.

Homogeneity assessment: The homogeneity of the material was assessed using purity assay by HPLC with UV detection on ten randomly selected 1-2 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

Safety: Treat as a hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

S. R. Davies

Dr Stephen R. Davies,
Team Leader,
Chemical Reference Materials, NMI.
23 September 2022

This report supersedes any issued prior to 23 September 2022.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

Characterisation Report:

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The certified purity value was obtained by mass balance from a combination of traditional analytical techniques, including HPLC with UV detection, thermogravimetric analysis, Karl Fischer analysis and ¹H NMR spectroscopy. The purity value is calculated as per Equation 1.

$$\text{Purity} = (100\% - I_{\text{ORG}}) \times (100\% - I_{\text{VOL}} - I_{\text{NVR}}) \quad \text{Equation 1}$$

I_{ORG} = Organic impurities of related structure, I_{VOL} = volatile impurities, I_{NVR} = non-volatile residue.

Supporting evidence is provided by elemental microanalysis.

HPLC:	Instrument:	Waters Model 1525 Binary pump, 717 plus auto sampler
	Column:	Altima C-18, 5 μm (4.6 mm \times 150 mm)
	Column oven:	Ambient
	Mobile Phase:	Acetonitrile/20 mM ammonium acetate pH 4.2 (55:45)
	Flow rate:	1.0 mL/min
	Detector:	Waters 2998 PDA operating at 225 nm
	Relative mass fraction of the main component:	
	Initial analysis:	Mean = 99.2%, s = 0.05% (10 sub samples in duplicate, July 2001)
	Re-analysis:	Mean = 99.5%, s = 0.05% (5 sub samples in duplicate, March 2011)
	Re-analysis:	Mean = 99.5%, s = 0.02% (5 sub samples in duplicate, January 2018)
	Re-analysis:	Mean = 99.4%, s = 0.01% (5 sub samples in duplicate, September 2022)
Karl Fischer analysis:		Moisture content \leq 0.2% mass fraction (2008, 2011, 2018 and 2022)
Thermogravimetric analysis:		Volatiles content < 0.1% and non-volatile residue < 0.2% mass fraction (August 2006)

Spectroscopic and other characterisation data

ESI-MS:	Instrument:	Finnigan MAT TSQ 700 with electrospray interface
	Operation:	Positive/Negative ion mode, direct infusion at 5 μ L/min
	Ionisation:	ESI spray voltage at 4.5 kV positive ion, 2.5 kV for negative ion
	Peak:	375, 377, 379 (main isotope peaks for $[M+H]^+$, positive ion mode) <i>m/z</i> 373, 375, 377 (main isotope peaks for $[M-H]^-$, negative ion mode) <i>m/z</i>
HRMS:	Found:	373.9457
	Calculated:	373.9450 for $C_{14}H_9^{35}Cl_3N_2O_2S$
IR:	Instrument:	FT-IR, Biorad WIN FTS40
	Range:	4000-400 cm^{-1} , KBr disc
	Peaks:	3203, 1578, 1445, 1390, 1327, 1255, 1166, 1141, 1044, 972, 909, 870, 773 cm^{-1}
1H NMR:	Instrument:	Bruker DMX-300
	Field strength:	300 MHz
	Solvent:	DMSO- d_6 (2.50 ppm)
	Spectral data:	δ 3.10 (3H, s), 3.33 (1H, br s), 6.77 (1H, dd), 7.29 (1H, t), 7.40 (1H, dd), 7.47 (1H, s), 7.93 (1H, s) ppm Ethyl acetate estimated at 0.6% mass fraction was observed in the 1H NMR
^{13}C NMR:	Instrument:	Bruker DMX-300
	Field strength:	75.5 MHz
	Solvent:	DMSO- d_6 (39.6 ppm)
	Spectral data:	δ 40.8, 108.6 (br), 116.3, 118.3 (br), 121.1, 121.8, 125.2, 129.1, 133.3, 138.0 (br), 139.0 (br), 146.8, 154.6, 158.1 ppm
Melting point:		198-199.5 $^{\circ}C$
Microanalysis:	Found:	C = 44.9%; H = 2.5%; N = 7.5%; Cl = 28.2% (June 2001)
	Calculated:	C = 44.8%; H = 2.4%; N = 7.5%; Cl = 28.3% (Calculated for $C_{14}H_9Cl_3N_2O_2S$)