

Australian Government

Department of Industry, Science and Resources

## National Measurement Institute



HCI

# CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

## NMIA D746b: Ecgonine hydrochloride

Report ID: D746b.2024.01

Chemical Formula: C<sub>9</sub>H<sub>15</sub>NO<sub>3</sub>.HCl

Molecular Weight: 221.7 g/mol

### **Certified value**

Batch No.	CAS No.	Purity (mass fraction)
08-D-09	5796-31-6	99.3 ± 0.7%

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit (k = 2).

IUPAC name: (1R,2R,3S,5S)-2-Carboxy-3-hydroxy-8-methyl-8-azoniabicyclo[3.2.1]octane chloride.

**Expiration of certification:** The property values are valid till 24 April 2029, five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** White powder prepared by synthesis, certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

**Intended use:** This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

Recommended storage: When not in use this material should be stored at or below 25 °C in a closed container in a dry, dark area.

**Metrological traceability:** The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. Quantitative NMR provides an independent direct measure of the mass fraction of the analyte of interest, calibrated with an internal standard certified for purity (mass fraction).

**Stability:** This material has demonstrated stability over a minimum period of ten years. The measurement uncertainty at the 95% coverage interval includes a stability component which has been estimated from annual stability trials.

The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by qNMR on five randomly selected 10-20 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Safety:** Treat as hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 27 June 2024

This report supersedes any issued prior to 22 June 2024.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website..

#### **Characterisation Report**

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The certified purity value was obtained by quantitative nuclear magnetic resonance (qNMR). The three-proton singlet at 2.8 ppm was measured against a certified internal standard of maleic acid.

Supporting evidence is provided by Karl Fischer analysis, <sup>1</sup>H NMR analysis and elemental microanalysis.

QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis: Re-analysis:	Bruker DMX 400 and 600 MHz $D_2O$ Potassium hydrogen maleate Purity = 99.3% (mass fraction, n = 4, s = 0.2%, August 2008) Purity = 99.5% (mass fraction, n = 5, s = 0.8%, September 2009)
QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis: Re-analysis:	Bruker Avance III-400 400 MHz $D_2O$ Maleic acid Purity = 99.1% (mass fraction, five samples in duplicate, s = 0.3%, January 2011) Purity = 99.1% (mass fraction, five samples in duplicate, s = 0.3%, November 2013)
QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis:	Bruker Avance III-500 500 MHz $D_2O$ Maleic acid Purity = 99.7% (mass fraction, five samples in duplicate, s = 0.2%, October 2016)
Karl Fischer analysis:		Moisture content 0.3% mass fraction (September 2008) Moisture content 0.2% mass fraction (January 2011 and November 2013) Moisture content $\leq$ 0.1% mass fraction (November 2016, September 2019 and April 2024)

### Spectroscopic and other characterisation data

GC-MS:		Agilent 6890/5973 DB-1701, 30 m × 0.25 mm l.D. × 0.25 $\mu$ m 180 °C (1 min), 4 °C/min to 200 °C, 6 °C/min to 275 °C (6.5 min) 230 °C 280 °C Helium, 1.3 mL/min Splitless e <i>bis</i> -trimethylsilyl derivative is reported along with the major peaks in the mass spectrum. as mass/charge ratios and (in brackets) as a percentage relative to the base peak.
	4.3 min:	329 (M <sup>+</sup> , 9), 314 (15), 212 (11), 147 (12), 96 (62), 83 (100), 73 (39) <i>m/z</i>
TLC:	Conditions:	Kieselgel 60F <sub>254</sub> . Methanol/conc. NH <sub>3</sub> (200/3) Single spot observed, $R_f = 0.27$ . Visualisation with ninhydrin.
IR:	Instrument: Range: Peaks:	Biorad FTS300MX FT-IR 4000-400cm <sup>-1</sup> , KBr powder 3267, 3091, 2964, 1709, 1431, 1351, 1227, 1140, 1041, 881, 607 cm <sup>-1</sup>
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker DMX600 600 MHz $D_2O$ (4.79) $\delta$ 1.98-2.19 (4H, m), 2.25-2.41 (2H, m), 2.78 (3H, s), 3.18 (1H, dd, $J = 2.3, 7.1$ Hz), 3.92 (1H, m), 4.10 (1H, d, $J = 7.3$ Hz), 4.41 (1H, m) ppm
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker DMX600 150 MHz D₂O δ 23.2, 24.0, 35.5, 38.9, 49.5, 60.8, 63.6, 64.6, 176.4 ppm
Melting point:		233 °C
Microanalysis:	Found: Calculated:	C = 49.1 %; H = 7.4 %; N = 6.3 % (August 2008) C = 48.8 %; H = 7.3 %; N = 6.3 % (Calculated for $C_9H_{15}NO_3.HCI$ )