

Australian Government

Department of Industry, Science and Resources

## National Measurement Institute



# CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

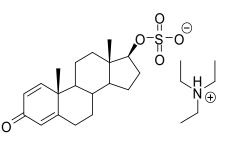
### NMIA D931: Boldenone sulfate, triethylamine salt

Report ID: D931.2024.01

Chemical Formula: C25H41NO5S

Molecular Weight: 467.7 g/mol

### **Certified value**



Batch No.	CAS No.	Purity (mass fraction)	
08-S-09	Not available	94.5 ± 0.6%	

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit (k = 2).

**Synonyms:** 1,4-Androstadiene-3-one-17β-ol sulfate, triethylamine salt.

**Expiration of certification:** The property values are valid till 4 September 2029, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** Off-white powder prepared by synthesis, and certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

Intended use: This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

**Recommended storage:** When not in use this material should be stored at or below 4 °C in a closed container in a dry, dark area.

**Metrological traceability:** The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. In the mass balance approach all impurities are quantified as a mass fraction and subtracted from 100%. Quantitative NMR provides an independent direct measure of the mass fraction of the analyte of interest, calibrated with an internal standard certified for purity (mass fraction).

**Stability**: This material has demonstrated stability over a minimum period of five years. The measurement uncertainty at the 95% confidence interval includes a stability component which has been estimated from annual stability trials. The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by HPLC with UV detection on ten randomly selected 1-2 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Safety:** Treat as a hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

Report ID: D931.2024.01 Product release date: 10 October 2008

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 10 September 2024

This report supersedes any issued prior to 10 September 2024.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The certified purity value was obtained by mass balance from a combination of traditional analytical techniques, including HPLC with UV detection, thermogravimetric analysis, Karl Fischer analysis and <sup>1</sup>H NMR spectroscopy. The purity value is calculated as per Equation 1.

Purity = (100 % - I<sub>ORG</sub>) x (100 % - I<sub>VOL</sub> - I<sub>NVR</sub>)

Equation 1

IORG = Organic impurities of related structure, IVOL = volatile impurities, INVR = non-volatile residue.

Supporting evidence is provided by LC-MS analysis, Q NMR analysis and elemental microanalysis.

HPLC:	Instrument: Column: Column oven: Mobile Phase:	Thermo Scientific UltiMate 3000 HPLC X-Bridge C-18, 5.0 μm (4.6 mm x 150 mm) 40 °C A= 20 mM NH4 <sup>+</sup> HCO2 <sup>-</sup> (aq) buffered to pH 10, B= Acetonitrile 0-18 min 22% B, 18-23 min 22-65% B, 23-25 min 65% B, 25-26 min 65-22% B,	
	Flow rate: Detector: Relative mass fractio Initial analysis:	26-35 min 22% B 1.0 mL/min, Gradient RS Photodiode Array Detector operating at 248 nm of the main component: Mean = 98.5%, s = 0.06% (3 sub samples in duplicate, September 2024)	
HPLC:	Instrument: Column: Column oven: Mobile Phase:	Waters HPLC X-Bridge C-18, 5.0 µm (4.6 mm x 150 mm) 40 °C A= 20 mM NH₄⁺ HCO₂⁻ (aq) buffered to pH 10, B=Methanol 0-15 min 22% B, 15-25 min 22-90% B, 25-30 min 90% B, 30-31 min 90-22% B, 31-46	
	Flow rate: Detector:	min 22% B 1.0 mL/min, Gradient Waters Photodiode Array Detector operating at 248 nm	
	Relative mass fractio Initial analysis: Re-analysis: Re-analysis:	n of the main component: Mean = 98.5%, s = 0.04% (6 sub samples in duplicate, June 2014) Mean = 98.7%, s = 0.02% (5 sub samples in duplicate, June 2017) Mean = 98.4%, s = 0.07% (5 sub samples in duplicate, May 2020)	
QNMR:	Instrument: Field strength: Solvent: Internal standard: Purity estimate:	Bruker DMX-600 600 MHz DMSO-d <sub>6</sub> (2.50 ppm) Dimethyl terephthalate Mean = 94.4%, s = 0.49% (5 sub samples in duplicate, April 2011)	
Thermogravimetric analysis:		Non volatile residue 0.4% mass fraction (May 2008) Non volatile residue 0.2% mass fraction (October 2010)	
Karl Fischer analysis:		Moisture content 2.6% mass fraction (September 2008) Moisture content 3.5% mass fraction (October 2009) Moisture content 3.8% mass fraction (October 2010) Moisture content 4.1% mass fraction (March 2014) Moisture content 4.5% mass fraction (June 2017) Moisture content 3.9% mass fraction (May 2020)	

Moisture content 4.2% mass fraction (August 2024)

#### Spectroscopic and other characterisation data

LC-MS:	Instrument: Column: Column temp: Solvent system: Flow rate: Sample prep: Injection volume: Ionisation mode: Capillary voltage: Source temp: Cone gas flow rate:	Waters 2695 (HPLC)/Micromass QuatroX-Bridge C-18, 150 mm × 4.6 mm I.D. × 5 $\mu$ m40 °C20 mM NH4 <sup>+</sup> HCO2 <sup>-</sup> (aq) buffered to pH 10, methanol0.2 mL/min2000 $\mu$ g/g in MeOH/20 mM NH4 <sup>+</sup> HCO2 <sup>-</sup> (aq) buffered to pH 10 (22:78)10 $\mu$ LElectrospray negative ion3.0 kVCone voltage:20 V130 °CDesolvation gas temperature:350 °C27 L/hrDesolvation gas flow rate:748 L/hr		
ESI-MS:	Instrument Operation: Ionisation: EM voltage: Cone voltage: Peak:	Micromass Quatro Micro Negative ion mode, direct infusion at 5 μL/min ESI spray voltage at 3.2 kV negative ion 500 V 20 V 365 (M-TEA <sup>+</sup> ) <sup>-</sup> <i>m/z</i>		
TLC:	Conditions:	Kieselgel 60F <sub>254</sub> . Dichloromethane / Methanol (9/1) Single, broad spot observed, $R_f$ = 0.23. Visualisation with UV at 254 nm		
IR:	Instrument: Range: Peaks:	Biorad FTS300MX FT-IR 4000-400 cm <sup>-1</sup> , KBr powder 3362, 2939, 2676, 2492, 1659, 1474, 1446, 1257, 1220, 1000, 804 cm <sup>-1</sup>		
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Gauss-400 400 MHz MeOH-d₄ (3.30 ppm) δ 0.91 (3H, s), 0.98-1.09 (3H, m), 1.20 (1H, m), 1.28 (3H, s), 1.31 (9H, t, <i>J</i> = 7.3 Hz), 1.40 (1H, m), 1.59-1.84 (5H, m), 2.01 (2H, m), 2.16 (1H, m), 2.39 (1H, m), 2.58 (1H, m), 3.21 (6H, q, <i>J</i> = 7.3 Hz), 4.21 (1H, dd, <i>J</i> = 8.1, 8.9 Hz), 6.05 (1H, m), 6.20 (1H, dd, <i>J</i> = 1.9, 10.1 Hz), 7.29 (1H, d, <i>J</i> = 10.1 Hz) ppm		
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Gauss-400 100 MHz MeOH-d₄ (49.0 ppm) δ 9.2, 12.1, 19.1, 23.5, 24.4, 29.1, 33.8, 34.5, 36.6, 37.7, 44.1, 45.4, 47.9, 50.9, 54.2, 87.7, 124.0, 127.5, 159.6, 173.5, 188.6 ppm		
Melting point:		108-112 °C		
Microanalysis:	Found: Calculated: Calculated:	C = 62.3 %; H = 8.7 %; N = 3.0% (September 2008) C = 64.2 %; H = 8.8 %; N = 3.0% (Calc. for $C_{25}H_{41}NO_5S$ ) C = 62.3 %; H = 8.9 %; N = 2.9% (Calc. for $C_{25}H_{41}NO_5S$ containing 0.8 molecule of moisture)		